14 October 1998

BAKER BOTTS U.P TRANSMITTAL LETTER TO THE UNITED STATES DESIGNATED/ELECTED OFFICE (DO/EO/US) CONCERNING A FILING UNDER 35.U.S.C. 371 INTERNATIONAL APPLICATION NO INTERNATIONAL APPLICATION NO INTERNATIONAL FILING DATE EXPRESS MAIL LABEL No EF321686401US 04/16/01 ATTORNEY'S DOCKET NO A32011-A-PCT-USA US APPLICATION NO INTERNATIONAL FILING DATE PRIORITY DATE CLAIMED

PCT/US99/23902 14 October 1999

TITLE OF INVENTION IMMOBILIZED SILVER IMMUNASSAY SYSTEM

APPLICANT(S) FOR DO/EO/US Antonio A. Garcia and Matthew R. Bonen

Applicant herewith submits to the United States Designated /Elected Office (DO/EO/US) the following items and other information:

- 1. This is a FIRST submission of items concerning a filing under 35 U.S.C. 371.
- 2. [] This is a SECOND or SUBSEQUENT submission of items concerning a filing under 35 U.S.C. 371.
- 3. This express request to begin national examination procedures (35 U.S.C. 371(f)) at any time rather than delay examination until the expiration of the applicable time limit set in 35 U.S.C. 371(b) and PCT Articles 22 and 39(I).
- 4. A proper Demand for International Preliminary Examination was made by the 19th month from the earliest claimed priority date.
- 5. A copy of the International Application as filed (35 U.S.C. 371(c)(2))
 - a. [] is transmitted herewith (required only if not transmitted by the International Bureau).
 - b. M has been transmitted by the International Bureau.
 - c. [] is not required, as the application was filed in the United States Receiving Office (RO/US).
- 6. [] A translation of the International Application into English (35 U.S.C. 371(c)(2)).
- 7. A copy of the International Search Report (PCT/ISA/210)
 - a. [] are transmitted herewith (required only if not transmitted by the International Bureau).
 - b. M have been transmitted by the International Bureau
 - c. [] have not been made; however, the time limit for making such amendments has NOT expired.
 - d. [] have not been made and will not be made.
- 8. [] A translation of the amendments to the claims under PCT Article 19 (35 U.S.C. 371(c)(3)).
- 9. [] An oath or declaration of the inventor(s) (35 U.S.C. 371(c)(4)).
- 10. [] A translation of the annexes to the International Preliminary Examination Report under PCT Article 36 (35 U.S.C. 371(c)(5)).

Items 11. to 16. below concern other document(s) or information included:

- 11. A copy of the International Preliminary Examination Report (PCT/IPEA/409)
- 12. An assignment document for recording. A separate cover sheet in compliance with 37 CFR 3.28 and 3.31 is included.
- 13. [] A FIRST preliminary amendment.
 - [] A SECOND or SUBSEQUENT preliminary amendment.
- 14. [] A substitute specification.
- 15. [] A change of power of attorney and/or address letter.
- 16. Other items or information:
 - a. a copy of the International Search Report (PCT/ISA/210)
 - b. A a copy of the International Preliminary Examination Report (PCT/IPEA/409)

INTERNATIONAL PROPERTIES DE 100 PCT/US99/1992/1892/1807	internationa 14 Octobe	L FILING DATE r 1999		PRIORITY DATE CLAIMED 14 October 1998	
17. The following fees are submitted:	_CALCULATIONS				
Basic National Fee (37 CFR 1.492(a)					
Neither international preliminary examina	. , . ,	FR 1.482)			
Nor international search fee (37 CFR 1.44 Report not prepared by the EPO or JPO (1	5(a)(2)) paid t	to USPTO and Internati	onal Search		
International preliminary examination fee International Search Report prepared by the	(37 CFR 1.48	32) not paid to USPTO	but 60.00		
International preliminary examination fee international search fee (37 CFR 1.445(a)(e (37 CFR 1.48	82) not paid to USPTO	but 10.00		
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International preliminary examination fee satisfied provisions of PCT Article 33(1)-(all claims		
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Claims	Number Filed	Number Extra	Rate	\$	
Total Claims	23 -20=	3	X \$ 18.00	\$ 54	
Independent Claims	8 -3=	5	X \$ 80.00	\$ 400	
Multiple dependent claim(s) (if applicable	le)		+ \$270.00	\$	
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c. 📝 The Commissioner is hereby author	rized to char	rge any additional fee	es which may be r	equired, or credit any	overpayment to
Deposit Account No. <u>02-4377</u> .	A copy of th	is sheet is enclosed.			
NOTE: Where an appropriate time lin (b)) must be filed and granted to restor				et, a petition to revi	ve (37 CFR 1.137(a) or
SEND ALL CORRESPONDENCE TO:				Marta	E Delsynne
BAKER BOTTS L.L.P.				Signature	
30 Rockefeller Plaza New York, New York 10112-4498				04/16/01	O
New 101k, New 101k 10112-4476				Date	
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PATENT

IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

Applicant

Garcia et al.

Serial No.

09/807,663

Filed

April 16, 2001

For

IMMOBILIZED SILVER IMMUNOASSAY SYSTEM

PRELIMINARY AMENDMENT

I hereby certify that this paper is being deposited with the United States Postal Service as First Class Mail in an envelope addressed to: Assistant Commissioner for Patents, Washington, D.C. 20231

May 4, 2001

Date of Deposit

Marta E. Delsignore

Attorney Name

Huri

Sionature

32.689

PTO Registration No.

May 4, 2001

Date of Signature

Assistant Commissioner for Patents Washington, D.C. 20231

Sir:

Preliminary to examination, please amend the above-identified patent application

as follows.

IN THE SPECIFICATION

Please insert on Page 1, before the first sentence the following:

--This application is a national stage application of PCT/US99/23902 which claims the benefit of priority of provisional applications 60/104,263 filed October 14, 1998 and 60/145,786 filed July 27, 1999. PCT/US99/23902 was published in English under publication number WO 00/21665 on April 20, 2000.--

Respectfully submitted,

Marta E. Delsignore

Patent Office Reg. No. 32,689

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IMMOBILIZED SILVER IMMUNOASSAY SYSTEM

Specification

Background of the Invention

Since the late 1950s, when the work of Rosalyn Yalow and Solomon Berson (Yalow et al., 1959, Nature (London) 184:1648; Yalow et al., 1960, J. Clin. Invest 39:1157) first illuminated the possibilities of immunoassays, immunodiagnostic techniques based on the specific interaction of antibody and antigen have become of paramount importance in the clinical, agricultural, food, veterinary, and environmental sectors. In 1996, it was estimated that the worldwide market of immunoassay products was \$10 billion in the clinical sector alone, and increasing at an annual rate of 10% (Deshpande, Enzyme Immunoassay: From Concept to Product Development, New York: Chapman & Hall, 1996). This market is driven by an ever-increasing desire for assays of greater sensitivity and specificity, at reasonable financial costs. It is in this environment that the specific and irreversible interaction between avidin or streptavidin and biotin has found use.

Streptavidin, a close relative of egg white avidin, is expressed in *Streptomyces avidinii* (Green, 1990, Methods in Enzymology 184:51), and both avidin and streptavidin exhibit an affinity for biotin on the order of 10¹⁵ M⁻¹. The streptavidin-biotin system has become a widely-used tool of molecular biology in such applications as affinity chromatography, cytometry, nucleic acid research, and diagnostics (Diamandis et al., 1991, Clin. Chem. 37:625; Wilchek et al., 1988, Anal. Bio Chem. 171:1). A common immunological procedure calls for the use of streptavidin-coated microtiter plates, which are used to capture either biotinylated antibodies or antigens. Since the assay is based on the interaction of streptavidin and biotin, universal kit-based assay formats are possible. These universal assays are also the basis of many automated immunological testing systems (Chan, ed., 1996, Immunoassay Automation: An Updated Guide to Systems, San Diego: Academic Press, 51-308). A format utilizing a microtiter-based enzyme-linked immunosorbent

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assay (ELISA) can measure a wide variety of analytes using a dual antibody or "sandwich" immunoassay. The choice of a streptavidin-coated solid support is made to overcome the limitations present in the direct antibody coating of polystyrene supports, which can result in unreliable or nonuniform coating of the solid support, in addition to the steric affects of binding upon the antibody. However, while streptavidin systems allow for universal ELISA kits and can improve assay sensitivity, coated plates can be costly.

Research examining the behavior of biotin has found that immobilized silver ions will bind biotinylated compounds both strongly and, in this case, reversibly. This research has been done with both immobilized metal affinity chromatography (IMAC) (Garcia et al., 1994, Reactive Polymers 23:249; Kim et al., 1995, Art. Cells, Blood Subs., and Immob. Biotech. 23:555; Miles et al., 1995, J. Chromatogr. A. 702:173) and paramagnetic particles (Ramirez-Vick, 1997, Ph.D. Dissertation, Arizona State University, Tempe, AZ). It has been discovered in accordance with the present invention that immobilized silver ions can be used in an immunoassay format to provide a sensitive and inexpensive universal assay.

Summary of the Invention

The present invention provides an immunoassay system comprising bioassay plates having silver immobilized thereon. The present invention further provides a method of making bioassay plates having silver immobilized thereon.

In another embodiment, the present invention provides a method for detecting an antigen or antibody, and a kit useful for the detection of an antigen or antibody.

An apparatus for providing activated bioassay plates is also provided by the present invention.

25 <u>Brief Description of the Drawings</u>

Fig. 1 is a graph depicting the sensitivity of a microtiter plate having silver immobilized thereon.

Fig. 2 is a schematic of an immunoassay utilizing microtiter plates having silver immobilized thereon, and biotinylated capture antibodies.

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Fig. 3 is a graph providing a kinetic analysis of a horseradish peroxidase immunoassay.

- Fig. 4 is a schematic of a checkerboard assay.
- Fig. 5 shows the results of a checkerboard assay.
- Fig. 6 is a schematic depicting the arrangement of activated and control wells of a microtiter plate.
 - Fig. 7 is a graph of immunoassay results comparing a streptavidin method with the silver method of the present invention.
 - Fig. 8 is a flow diagram describing the apparatus of the present invention.
- Fig. 9 is a side view of the apparatus of the invention.
 - Fig. 10 is a diagram of the liquid handling system of the invention.
 - Figs. 11a-11c are diagrams of the liquid transfer manifold of the invention. Fig. 11a is a side view of the reagent addition stage and vacuum stage; Fig. 11b is a front view of the reagent addition stage; Fig. 11c is a front view of the vacuum stage.

15 <u>Detailed Description of the Invention</u>

The present invention provides an immunoassay system comprising bioassay plates having silver, in particular silver ions, immobilized thereon. The invention further provides methods of making and using such bioassay plates. The bioassay plates and immunoassay of the present invention are useful for the detection of antibodies and antigens, and provide cost and sensitivity advantages relative to the streptavidin-coated bioassay plates of the prior art.

The bioassay plates used in the present invention are microwell, or microtiter, plates known in the art for immunoassays, and are commercially available. Conventional microwell plates are 96-well microplates having wells arranged on an 8 x 12 matrix on 9 mm centers. Each well holds approximately 300 microliters. 384-well plates are also available, in which the wells are arranged in a 16 x 24 matrix on a 4.5 mm center, with each wells having a brim volume of approximately 80 microliters. Well plates defined by larger matrices, e.g. 1536 well plates, are also available. The number and configuration of the wells are not critical to the present

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invention, and are used by way of example only. The bioassay plates used in accordance with the present invention are plastic, and preferably polystyrene.

Bioassay plates having silver ions immobilized thereon are made by a method comprising functionalizing a multi-well bioassay plate to provide an amine-containing bioassay plate, adding polymerized glutaraldehyde to the wells of the plate for a time and under conditions whereby the amines are activated by glutaraldehyde, rinsing the plates with an aqueous solution, adding thiourea to the wells of the plate for a time and under conditions whereby the thiourea is reacted with a glutaraldehyde moiety of the glutaraldehyde-activated bioassay plate, rinsing the plate with an aqueous solution, and contacting the plate with silver ions for a time and under conditions whereby the silver ions are immobilized on said plate.

The bioassay plate may be constructed of any material that can be functionalized to contain an amine group. In a preferred embodiment, the multi-well bioassay plate is a plastic multi-well bioassay plate. For example, the plate may be made of polystyrene, polyethylene, polypropylene, or other primary polymers or composite resins. Polystyrene is particularly preferred. Methods for functionalizing these materials to contain an amine group are known in the art. For example, a polystyrene bioassay plate can be functionalized to contain an amine group by methods known in the art and disclosed for example in Immobilized Affinity Ligand Techniques, Hermanson et al., eds., San Diego: Academic Press, 1992, the disclosure of which is incorporated herein by reference. Aminated polystyrene bioassay plates are also commercially available, for example from Corning (Corning, NY), NUNC (Denmark) and Micro Membranes (Newark, NJ). Preferably the plate is amidated or aminated to contain from about 1 x 10¹³ to 1 x 10¹⁴ amine sites per cm².

Polymerized glutaraldehyde may be prepared by allowing glutaraldehyde (25 wt %) to polymerize, for example for from 1 to 36 hours at from 23 °C to 70 °C. In a preferred embodiment polymerization is at 70 °C for about 24 hours. The polymerized glutaraldehyde is added to the wells of the plate and incubated under conditions whereby a glutaraldehyde-activated plate is produced, for example for from 1 to 36 hours at from 23 °C to 70 °C. In a preferred embodiment, incubation is at 35 to 50 °C for 1 to 24 hours, and more preferably at about 37 °C for about 24 hours. The

plate is rinsed with an aqueous solution, for example deionized water, to remove unreacted glutaraldehyde. The wells of the plate are then filled with thiourea, for example from 0.01M to 1M solution, and preferably a 1M solution, under conditions suitable for reaction with the glutaraldehyde moiety, for example for from 1 to 36 hours at from 23°C to 70°C. In a preferred embodiment at, incubation is for 1 to 24 hours at 35 to 50°C, and more preferably about 24 hours at about 37°C. The plate is rinsed with an aqueous solution, for example deionized water, to remove unreacted thiourea.

Silver ions, preferably in the form of silver nitrate, are then added to the plate under conditions whereby silver ions are immobilized on the plate, for example for from 1 to 36 hours at from 23°C to 70°C. In a preferred embodiment, incubation is for about 24 hours at about 37°C. The plates are then rinsed with an aqueous solution, for example deionized water, and may be stored until use, preferably in an opaque sleeve.

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The bioassay plates having silver ions immobilized thereon are useful in a method for detecting an antigen or an antibody. It has been discovered in accordance with the present invention that the silver ions immobilized on bioassay plates are capable of strong binding to biotinylated antibodies and antigens. Accordingly, the plates of the invention may be used in standard enzyme-linked immunosorbent assays (ELISAs). For example, a bioassay plate having silver ions immobilized thereon is incubated with biotinylated antibody to provide a bioassay plate having the antibody immobilized thereon. After washing with an aqueous solution, the plate is incubated with a solution containing the cognate antigen under conditions whereby the antigen binds to the immobilized antibody, followed by another washing step. The antigen is then detected, for example by subsequent incubation with a labeled antibody having specificity for the antigen. Detectable labels for antibodies are known in the art and include radiolabels, fluorescent tags, and enzyme conjugates. In a preferred embodiment, the aqueous solution contains deionized water and Tween (polyoxyethylene sorbitan monolaurate).

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An antibody may be detected using the plates of the present invention in an indirect ELISA assay. For example, a bioassay plate having silver ions immobilized

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thereon is incubated with a biotinylated antigen to provide a bioassay plate having the antigen immobilized thereon. After washing with an aqueous solution, the plate is incubated with a solution containing the cognate primary antibody under conditions whereby said antibody binds to the immobilized antigen. After incubation and washing, a labeled secondary antibody is added and incubated under conditions whereby it binds to the primary antibody. After washing, the secondary antibody is detected, wherein detection thereof indicates the presence of the primary antibody. Detectable labels for antigens are known in the art and include radiolabels, fluorescent tags, and enzyme conjugates. In a preferred embodiment, the aqueous solution contains deionized water and Tween.

Conditions for biotinylating antibodies and antigens are well known in the art and disclosed, for example, by Bayer et al., "Protein Biotinylation" (1990) Methods in Enzymology 184:138 and O'Shannessy "Antibodies Biotinylated via Sugar Moieties" (1990) Methods in Enzymology 184:162, the disclosures of which are incorporated herein by reference. Conditions for performing ELISAs are well-known in the art and disclosed, for example, by Harlowe, et al., (1988) Antibodies: A Laboratory Manual, Cold Spring Harbor Laboratory Press, Cold Spring Harbor, NY, the disclosure of which is incorporated herein by reference.

The present invention further provides a kit useful for the detection of an antigen or antibody. The kit comprises, in a first container, a bioassay plate having silver ions immobilized thereon. In a preferred embodiment, the bioassay plate is a polystyrene multi-well plate. The kit may optionally contain a second container containing a biotinylated antibody or a biotinylated antigen. The kit may optionally contain a third container containing labeled antibody, when the second container contains a biotinylated antibody, or a labeled secondary antibody, when the second container contains a biotinylated antigen.

The present invention further provides an apparatus useful for the automated production of microplates having modified surface chemistry. As shown by the flow diagram in Fig. 8, the apparatus provides for filling the wells of a microplate with a reagent in an addition/withdrawal chamber; conveying the microplate to an incubation chamber in which the microplate is sealed, heated and agitated, and unsealed;

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conveying the microplate to the addition/withdrawal chamber for evacuation of reagent, washing, and addition of a second reagent; conveying the microplate to the incubation chamber for sealing, heating and agitation, and unsealing; conveying the microplate to the addition/withdrawal chamber for evacuation of reagent and washing; followed by subsequent cycles of reagent addition and incubation, or conveyance of the microplate out of the machine.

In a preferred embodiment, and with reference to Figs. 9, 10, 11a, 11b and 11c, the apparatus comprises a housing having disposed therein a reagent addition/withdrawal chamber (1) and an incubation chamber (2). Microplates are conveyed into and between the chambers by means of a plate holder (3) movable horizontally by a plate holder track (4). Reagents and wash solution are provided in storage containers (5) connected by reagent lines (13) to the dispense portion of a manifold (6) which delivers reagent and wash solution by dispense lines (7) by means of a liquid pump (11). After reagent addition, microplates positioned on the plate holder (3) are conveyed via the plate holder track (4) into the incubation chamber (2). The microplate is sealed by a non-reactive sealing plate (13) delivered vertically. The incubation chamber further provides a means for heating and agitating the microplate (14). After a time predetermined by the user, the microplate is conveyed to the reagent addition/withdrawal chamber (1) via plate holder track (4). Spent reagent is removed through aspirator lines (8) and withdrawn by the aspirate portion of the same manifold (6) by means of a vacuum pump (9) through waste lines (12) to a waste container (10). Wash solution is added through the dispense portion of the manifold (6) which delivers wash solution through dispense lines (7). Wash solution is removed through aspirator lines (8) and withdrawn by the aspirate portion of manifold (6) by means of a vacuum pump (9) through waste lines (12) to a waste container (10).

The foregoing steps are carried out in an automatic programmed manner under the control of electronic circuitry contained in the housing.

All references cited herein are incorporated in their entirety.

The following examples serve to further illustrate the present invention.

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Example 1

A polystyrene 96 well microtiter plate aminated to provide approximately 2 x 10¹³ active amine sites per cm² was obtained from Corning (Corning, NY). Glutaraldehyde (25 wt %) which had been allowed to polymerize at 70°C for 24 hours was added to each well of the microplate, which was then incubated at 37°C for 24 hours to facilitate plate activation. The plate was then rinsed with deionized water and wells filled with a 1M solution of thiourea, followed by an additional 24 hour period of incubation at 37°C. After another rinsing of the plate, a 1M solution of silver nitrate was allowed to contact the plate during another 24 hour incubation at 37°C. The plate was then rinsed extensively.

The biotin-binding capability of such an immobilized silver microtiter plate was tested as follows. A complete plate was assembled from stripwells, using alternating strips of unactivated and silver-containing wells, where the silver-containing strips were in the odd-numbered rows and the unactivated strips were in the even-numbered rows. The test consisted of the binding of biotinylated horseradish peroxidase (bHRPO), which was detected using the chromogenic reaction of 2,2'-azino-bis(3-ethylbenzthiazoline-6-sulfonic acid) diammonium salt (ABTS®). This was a one-step, unamplified assay; 150 µl of the bHRPO solution was added to the wells using doubling dilutions, and allowed to bind for one hour. Following this, the plates were rinsed with deionized water and the ABTS® in citrate buffer was added. The developed color in the wells was read after one hour at room temperature using a BioRad Benchmark Microplate Reader set at 415 nm. The plate setup and moles of bHRPO corresponding to each dilution are shown in Tables I and II, respectively.

TABLE I

	1 2	3 4	5 6) 7 8	3 9 10	11 12
Α	no bHRPO	no bHRPO	no bHRPO	no bHRPO	no bHRPO	no bHRPO
В	no bHRPO	1/1	1 / 64	1 / 4096	1 / 262144	no bHRPO
5 c	no bHRPO	1/2	1 / 128	1 / 8192	1 / 524288	no bHRPO
D	no bHRPO	1/4	1 / 256	1 / 16384	1 / 1048576	no bHRPO
E	no bHRPO/ABTS	1/8	1/512	1 / 32768	1 / 2097152	no bHRPO/ABTS
F	no bHRPO/ABTS	1/16	1 / 1024	1 / 65536	1 / 4194304	no bHRPO/ABTS
G	no bHRPO/ABTS	1 / 32	1 / 2048	1 / 131072	1 / 8388608	no bHRPO/ABTS
10 н	no bHRPO/ABTS	no bHRPO/ABTS	no bHRPO/ABTS	no bHRPO/ABTS	no bHRPO/ABTS	no bHRPO/ABTS

TABLE II

	Dilution	Grams bHRPO	Moles bHRPO
	1 / 1	2.40E-06	5.33E-11
	1/2	1.20E-06	2.67E-11
15	1 / 4	6.00E-07	1.33E-11
	1 / 8	3.00E-07	6.67E-12
	1 / 16	1.50E-07	3.33E-12
	1/32	7.50E-08	1.67E-12
	1 / 64	3.75E-08	8.33E-13
20	1 / 128	1.88E-08	4.17E-13
	1 / 256	9.38E-09	2.08E-13
	1 / 512	4.69E-09	1.04E-13
	1 / 1024	2.34E-09	5.21E-14
	1 / 2048	1.17E-09	2.60E-14
25	1 / 4096	5.86E-10	1.30E-14
	1 / 8192	2.93E-10	6.51E-15
	1 / 16384	1.46E-10	3.26E-15

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TABLE II

Dilution	Grams bHRPO	Moles bHRPO
1 / 32768	7.32E-11	1.63E-15
1 / 65536	3.66E-11	8.14E-16
1 / 1E+05	1.83E-11	4.07E-16
1 / 3E+05	9.16E-12	2.03E-16
1 / 5E+05	4.58E-12	1.02E-16
1 / 1E+06	2.29E-12	5.09E-17
1 / 2E+06	1.14E-12	2.54E-17
1 / 4E+06	5.72E-12	1.27E-17
1 / 8E+06	2.85E-13	6.36E-18

After applying statistical curve fitting techniques, a concentration activity curve was generated and is shown in Figure 1. Results are presented in terms of moles of bHRPO. As demonstrated therein, the detection limit of this system approaches femtomolar levels, even using an unamplified system. The typical Gaussian response curve of ELISA is also seen in Figure 1, which further demonstrates that the maximum sensitivity of this assay occurs at bHRPO levels of 2 x 10⁻¹³ moles.

Example II

Example I demonstrated that immobilized silver microtiter plates are capable of binding biotin and a biotinylated antigen. This example demonstrates that biotinylated capture antibodies can be bound to immobilized silver microtiter plates and used for antigen capture.

Immobilized silver microtiter plates were prepared as described in Example I. The plates were incubated with biotinylated anti-peroxidase antibodies (Jackson Immunological) at an antibody concentration of 1.2 mg/ml diluted 1:100, with addition of 150 microliters to each well for an hour. After washing the plates with 50 mM phosphate buffer with 0.1% v/v Tween 20 detergent (Phosphate/Tween buffer), the enzyme horseradish peroxidase (Sigma) was added, in the amounts shown in

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Table III, for one hour incubation. Phosphate/Tween buffer was used as the dilution buffer in the assay.

The results of eight assays of the type shown in Figure 2 are presented in Figure 3, along with 2σ error limits. The data for the immunoassays is shown in Table III, using the eight simultaneous assay replications for a basic statistical analysis. Figure 3 uses kinetic rates to determine the detection of enzyme, which allows for a linear fit of the data when presented on a semi-log plot using the dilution number of the initial enzyme solution. At the initial dilutions, the margin of error is rather high, due to the high amounts of enzyme in the initial solution (2.56 x 10^{-11} moles per well) producing extremely rapid kinetics. As the enzyme goes through doubling dilutions, the kinetics are measured more easily and the precision of the assay improves at dilution number $16 (1.6 \times 10^{-12} \text{ moles per well})$. The detection limit of this assay is in the region of dilution number 1000 (0.25 femtomoles), which approaches the theoretical detection limit of the enzyme substrate system being used. (Deshpande, Enzyme Immunoassays: From Concept to Product Development, New York, Chapman & Hall (1996), 1-422.

TABLE III

		Kin	etic Immunoassay	/ Data	
	Dilution	HRPO	HRPO	Kinetic	
20	Number	(g/well)	(mol/well)	Rate	2σ Limits
	1	1.13E-06	2.56E-11	927.65	286.73
	2	5.63E-07	1.28E-11	773.66	253.20
	4	2.81E-07	6.40E-12	666.47	95.75
	8	1.41E-07	3.20E-12	598.88	97.22
25	16	7.03E-08	1.60E-12	551.07	61.58
	32	3.52E-08	8.00E-13	452.40	25.79
	64	1.76E-08	4.00E-13	361.72	24.07
	128	8.79E-09	2.00E-13	277.00	20.58
	256	4.39E-09	1.00E-13	193.03	27.47

TABLE III

	Kinetic Immunoassay Data					
Dilution	HRPO	HRPO	Kinetic			
Number	(g/well)	(mol/well)	Rate	2σ Limits		
512	2.20E-09	5.00E-14	128.36	49.72		
1024	1.10E-09	2.50E-14	80.85	53.07		
2048	5.49E-10	1.25E-14	65.28	62.02		
4096	2.75E-10	6.25E-15	60.26	61.08		
8192	1.37E-10	3.12E-15	62.54	60.02		
16384	6.87E-11	1.56E-15	52.25	78.54		
32768	3.43E-11	7.81E-16	46.92	83.12		
65536	1.72E-11	3.91E-16	36.93	69.52		
131072	8.58E-12	1.95E-16	40.52	69.90		

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The basis of the foregoing is the assumption that the enzyme binds only to the antibodies, which specifically select it from the test solution. In order for this to be valid, background binding of the enzyme to the plate must be at insignificant levels. In order to determine this, a checkerboard assay was performed (Deshpande, supra). The assay arrangement is shown in Figure 4. When performed, the initial concentrations of the antibody and the enzyme were the same as described above, as were all buffers and incubation periods. The checkerboard assay, which is primarily visual, is a means to determine the effect of the antibody concentration and the enzyme concentration on the assay results. As seen in Figure 5, the enzyme only binds when antibody is present in the well. As antibody concentrations decrease horizontally across the plate, enzyme binding falls rapidly, as shown by the absence of color on the right side of the plate, even at the extreme enzyme concentrations introduced into row A.

Example III

The foregoing examples demonstrate that immobilized silver is capable of binding biotin in the bHRPO assay and that effective immobilization of capture antibodies in the immobilized silver microplate wells is possible. In the present example, the silver ion immunoassay format is then compared to the current streptavidin technology used to bind biotinylated antibodies. Plates coated with streptavidin (Xenopore, XPS 010 00) were obtained for this purpose. These plates represent the best of the currently available products since they feature a covalent linkage between the streptavidin and the plate surface, and are also blocked with a proprietary nonbiotinylated protein to inhibit background binding. No blocking agents are used on the silver plates. A side by side comparison with an immobilized silver plate prepared according to Example I was performed as follows.

Both the streptavidin plate and the silver plate were hydrated and rinsed with 50 mM pH 7 phosphate buffer. The wells in the plates were filled to capacity with buffer and allowed to stand at room temperature for 10 minutes. The plates were then rinsed twice with the same buffer.

Biotinylated anti-peroxidase antibodies (Jackson Immunological) were used to coat the plates. In this assay, 150 µl of antibody solution (1.2 mg/ml) diluted to 1:100 was added to the wells in the odd numbered columns. The even numbered columns were used as control wells, and were filled with 150 µl of buffer. This arrangement of activated and control wells is shown in Figure 6. The buffer used to dilute the antibodies and fill the control wells was 50 mM pH 7 phosphate buffer with 0.1% v/v Tween 20 Phosphate/Tween) added to inhibit any hydrophobic binding in the plates. The plates were then covered and allowed to stand at room temperature for 1.5 hours.

Both plates were thoroughly washed using a Bio-Rad Plate Washer filled with Phosphate/Tween buffer as the wash buffer. Horseradish peroxidase (Sigma) was used as the antigen in this test. A solution of peroxidase was created by adding 8 x 10^{-4} g of the enzyme to 10 ml of Phosphate/Tween. 1 ml of this solution was then added to 9 ml of buffer. 300 μ l of the diluted enzyme solution was added to wells A1, A2, A5, A6, A9, and A10. The amount of enzyme added to these primary wells is shown in Table IV. The remainder of the wells were filled with 150 μ l of

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Phosphate/Tween. For Test 1, 150 μ l of solution was withdrawn from A1 and A2 and diluted into B1 and B2. This procedure was repeated down the plate, with the solution from H1 and H2 carried to wells A3 and A4. Tests 2 and 3 were performed in a similar fashion. After all of these doubling dilutions were complete, the plates were covered and allowed to stand at room temperature for 1 hour.

For enzyme detection the plates were first washed using the automated plate washer with Phosphate/Tween. Then, 150 µl of an ABTS solution was added to each well in the plate, and the rate of formation of the colored product read at 415 nm in a Bio-Rad Benchmark Microplate Reader. Readings took place every 15 seconds for 5 minutes. The ABTS solution was made by adding 17 mg of ABTS to 100 ml of 50 mM pH 5 citrate buffer. Immediately before use, 100 µl of 3% hydrogen peroxide was added to the solution to catalyze the enzymatic reaction.

The raw data for these experiments is presented in Table IV and Table V. Using the average of the normalized tests, a plot comparing the streptavidin plate to the silver plate was prepared. The data points shown on the graph in Figs. 7 are the average values of the normalized tests shown in Tables IV and V. The exception to this are the data points taken at Dilution 1, which were replaced by the values representing the asymptotic approach of the immunoassays as determined from the preceding points. From this, the maximum kinetic rate achievable in the silver plate is 375, versus 150 in the streptavidin plate. Due to the approach to a final value at high levels of enzyme, this difference is due to the amount of capture antibody on the plate. The silver plate binds more functional capture antibody, and thus is able to bind more enzyme when excess enzyme is present. Assuming that each antibody captures an average of 1.5 enzyme molecules, it is possible to estimate that the silver plate has approximately twice as many biotinylated antibody binding sites available. The ultimate detection limits of both plates is the same, however, and approaches femtomolar levels. This is the limit of the enzyme/substrate detection system being used.

TABLE IV

Silver Plate Immunoassay Data

•	Dilimin	t/DDO					77		T	T	T . 3	<u> </u>
	Dilution Number	HRPO (mol/well)	Tes	<u>st 1</u>	<u>Tes</u>	<u> </u>	Tes	<u>st 5</u>	Test 1	Test 2	Test 3	Average
		(incl. item)	Antibody	Control	Antibody	Control	Antibody	Control	Normalized	Normalized	Normalized	
5	1	2 73E-11	2 83E+02	2 52E+01	2 38E+02	1 83E+01	2 90E+02	2 55E+01	2 58E+02	2 20E+02	2 65E+02	2 47E+02
	2	1 37E-11	3 87E+02	1 58E+0}	3 46E+02	1 15E+01	3 94E+02	1 73E+01	3 71E+02	3 34E+02	3 76E+02	3 61E+02
	4	6 83E-12	3 67E+02	9 82E+00	3 76E+02	7 55E+00	3 81E+02	1 22E+01	3 58E+02	3 68E+02	3 69E+02	3 65E+02
	8	3 41E-12	3 36E+02	6 04E+00	3 44E+02	4 20E+00	3 63E+02	8 71E+00	3 30E+02	3 40E+02	3 54E+02	3 41E+02
	16	1 71E-12	3 07E+02	5 74E+00	2 99E+02	3 37E+00	3 43E+02	7 77E+00	3.01E+02	2 96E+02	3 36E+02	3 11E+02
10	32	8 53E-13	2 67E+02	1 50E+01	2 70E+02	1 77E+00	2 90E+02	7 32E+00	2 52E+02	2 68E+02	2 83E+02	2 68E+02
	64	4 27E-13	2 18E+02	2 83E+00	2 16E+02	2 11E+00	2.28E+02	5 53E+00	2 15E+02	2 14E+02	2 22E+02	2 17E+02
	128	2 13E-13	1 61E+02	1 49E+00	1 53E+02	2 30E+00	1 50E+02	7 92E+00	1 60+02	1 51E+02	1 42E+02	1 51E+02
	256	1 07E-13	7 38E+01	1 04E+00	7 18E+01	4 99E+00	7 39E+01	2 02E+00	7 27E+01	6 68E+01	7 18E+01	7 05E+01
	512	5 33E-14	5 20E+01	1 35E+00	4 73E+01	2 57E+00	5 37E+01	2 32E+00	5.06E+01	4 48E+01	5 14E+01	4 89E+01
15	1024	2 67E-14	2 81E+01	1 49E+00	2 33E+01	4 83E+00	2 75E+01	3.26E+00	2 67E+01	1 85E+01	2 43E+01	2 31+01
	2048	1 33E-14	1 20E+01	1 96E+00	1 34E+01	5 77E+00	1 43E+01	4 13E+00	1 00E+01	7 68E+00	1 02E+01	9 30E+00
	4096	6 67E-15	6 66E+00	1 91E+00	I 05E+01	8 35E+00	8 61E+00	5 30E+00	4 75E+00	2 15 E +00	3 31E+00	3 41E+00
	8192	3 33E-15	4 73E+00	1 49E+00	1 03E+01	8 84E+00	3 03E+00	4 65E+00	3 24E+00	1 43E+00	-1 62E+00	1 02E+00
	16384	1 67E-15	2 93E+00	2 41E+00	7 07E+00	6 65E+00	6 78E+00	6 25E+00	5.22E-01	4 28E-01	5 25E-01	4 92E-01
20	32768	8 33E-16	2 12E+00	3 86E+00	5 24E+00	5 19E+00	6 95E+00	8 48E+00	-1 74E+00	4 82E-02	-1 53E+00	-1 07E+00

TABLE V

Streptavidin Plate Immunoassay Data

Dilution Number	HRPO (mol/well)	Ŀ	est I	Te	st 2	<u>Te</u>	st 3	Test 1	Test 2	Test 3	Average
Namber	(monwert)	Antibody	Control	Antibody	Control	Antibody	Control	Normalized	Normalized	Normalized	
5	2 73E-11	1 36E+02	6 50E+00	1 33E+02	8 55E+00	1 29E+02	5 50E+00	1 29E+02	1 24E+02	1 24E÷02	1 26E+02
2	1 37E-11	1 74E+02	8 61E+00	1 24E+02	3 46E+00	1 75E+02	7 26E+00	1 65E+02	1 20E+02	I 68E+02	1 51E+02
4	6 83E-12	1 74E+02	1 96E+00 _	1 51E+02	2 29E+00	I 49E+02	2 06E+01	1 72E+02	1 49E+02	1 28E+02	1 50E+02
8	3 41E-12	1 62E+02	1 47E-01	1 34E+02	1 02E+00	1 38E+02	4 83E-01	1 62E+02	1 33E+02	1 38E+02	1 44E+02
16	1 71E-12	1 62E+02	-2 36E-01	1 40E+02	3 03E-02	1 30E+02	-2 53E-01	1 62E+02	1 40E+02	1 30E+02	1 44E+02
10	8 53E-13	1 49E+02	-8 89E-01	1 16E+02	6 89E-01	1 31E+02	-1 41E+00	1 50E+02	1 16E+02	1 33E+02	1 33E+02
64	4 27E-13	1 23E+02	-6 99E-01	1 28E+02	3 23E-01	1 05E+02	-1 26E+00	1 24E+02	1 28E+02	1 06E+02	1 19E+02
128	2 13E-13	1 04E+02	-1 47E-01	1 01E+02	-8 88E-01	8 99E+01	-2 05E+00	1 04E+02	1 02E+02	9 20E+01	9 93E+01
256	1 07E-13	3 86E+01	8 42E-01	5 01E+01	1 06E+00	4 32E+0	5 47E-01	3 78E+01	491E+01	4 26E+01	4.32E+01
512	5 33E-14	3 95E+01	1 37E+00	3 31E+01	7 59E-02	4 04E+01	6 97E-02	3 81E+01	3 31E+01	4 03E+01	3 72E+01
$15 b_{24}$	2 67E-14	2 45E+01	1 38E+00	2 02E+01	-1 67E-01	2 45E+01	5 45E-01	2 31E+01	2 03E+01	2 40E+01	2 25E+01
2048	1 33E-14	1 29E+01	6 25E-01	9 56E+00	-8 19E-01	1.22E+01	1 06E-01	1 23E+01	1 04E+01	1 21E+01	1 16E+01
4096	6 67E-15	5 97E+00	-3 87E- 01	4 76E+00	-1 42E+00	6 07E+00	-3 69E-01	6 36E+00	6 18E+00	6 44E+00	6 33E+00
8192	3 33E-15	2 75E+00	-3 01E-01	1 47E+00	-2 08E+00	3 18E+00	-6 08E-02	3 05E+00	3 55E+00	3 24E+00	3 28E+00
16384	1 67E-15	1 25E+00	-9 64E-01	1 64E-01	-1 36E+00	-2 02E+00	-9 85E-01	2 21E+00	1 52E+00	-1 04E+00	8 97E-01
20768	8 33E-16	1 07E+00	-1 22E+00	-2 82E-01	-1 87E+00	-9 76E-01	-2 00E+00	2 29E+00	1 58E+00	1 02E+00	1 63E+00

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<u>Claims</u>

1.		A bioassay plate having silver ions immobilized thereon.
2.		The bioassay plate of Claim 1 wherein said plate is a polystyrene plate.
3.		The bioassay plate of Claim 2 wherein said plate is a multi-well plate.
4.		The bioassay plate of Claim 2 wherein said plate is a 96-well microplate.
5.		A multi-well bioassay plate having silver ions immobilized thereon made by a method comprising:
	a)	functionalizing a multi-well bioassay plate to provide an amine - containing bioassay plate;
	b)	adding a polymerized glutaraldehyde to the wells of said plate and maintaining for a time and under conditions to provide a glutaraldehyde - activated bioassay plate;
	c)	rinsing said plate with an aqueous solution;
	d)	adding thiourea to the wells of said plate and maintaining for a time
		and under conditions whereby the thiourea reacts with the
		glutaraldehyde moiety of said glutaraldehyde-activated bioassay plate;
	e)	rinsing said plate with an aqueous solution; and
	f)	contacting said plate with silver ions for a time sufficient to immobilize said silver ions on said plate.
6.		The multi-well bioassay plate of Claim 5 wherein said polymerized glutaraldehyde is prepared by allowing 25 wt % glutaraldehyde to

polymerize at about 70°C for about 24 hours.

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- 7. The multi-well bioassay plate of Claim 5 wherein said polymerized glutaraldehyde is maintained in the wells of said plate at about 37°C for about 24 hours. 8. The multi-well bioassay plate of Claim 5 wherein said thiourea is maintained in the wells of said plate at about 37°C for about 24 hours. 9. The multi-well bioassay plate of Claim 5 wherein said silver ions are added to said plate in the form of silver nitrate. 10. A method of making a multi-well bioassay plate having silver ions immobilized thereon comprising the steps of: functionalizing a multi-well bioassay plate to provide an aminea) containing bioassay plate; adding a polymerized glutaraldehyde to the wells of said plate and b) maintaining for a time and under conditions to provide a glutaraldehyde - activated bioassay plate; c) rinsing said plate with an aqueous solution; d) adding thiourea to the wells of said plate and maintaining for a time and under conditions whereby the thiourea reacts with the glutaraldehyde moiety of said glutaraldehyde activated bioassay plate; rinsing said plate with an aqueous solution; and e) f) contacting said plate with silver ions for a time sufficient to immobilize said silver ions on said plate. 11. The method of Claim 10 wherein said polymerized glutaraldehyde is prepared by allowing 25 wt % glutaraldehyde to polymerize at about 70°C for about 24 hours.
- 25 12. The method of Claim 10 wherein said polymerized glutaraldehyde is maintained in the wells of said plate at about 37°C for about 24 hours.

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e)

13. The method of Claim 10 wherein said thiourea is maintained in the wells of said plate at about 37°C for about 24 hours. 14. The method of Claim 10 wherein said silver ions are added to said plate in the form of silver nitrate. 15. A method for detecting an antigen comprising the steps of: a) incubating a multi-well bioassay plate having silver ions immobilized thereon with a biotinylated antibody having specificity for said antigen to provide a bioassay plate having said antibody immobilized thereon; b) incubating said plate with a solution containing said antigen; c) washing said plate with an aqueous solution; d) incubating said plate with a labeled antibody having specificity for said antigen; e) washing said plate with an aqueous solution; and f) detecting said label, wherein detection of said label is indicative of the presence of said antigen. 16. A method for detecting a first antibody comprising the steps of: a) incubating a multi-well bioassay plate having silver ions immobilized thereon with a biotinylated antigen that is reactive with said first antibody to provide a bioassay plate having said antigen immobilized thereon; b) incubating said plate with an aqueous solution containing said first antibody; washing said plate with an aqueous solution; c) incubating said plate with an aqueous solution containing a labeled d)

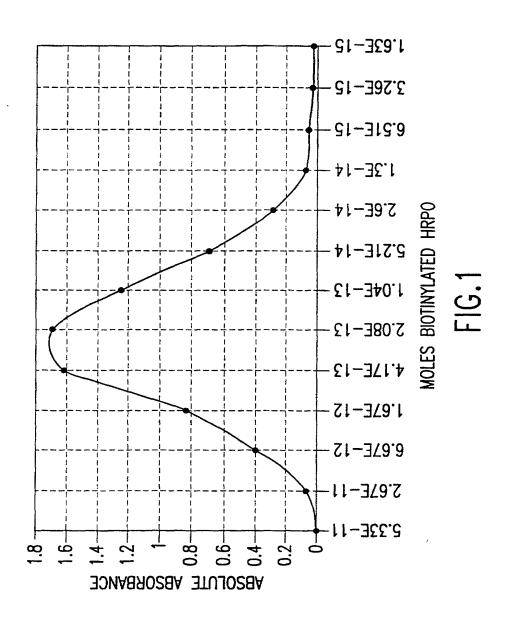
second antibody that binds to said first antibody;

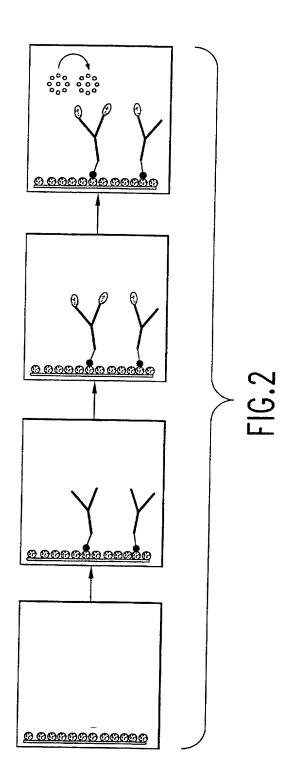
washing said plate with an aqueous solution; and

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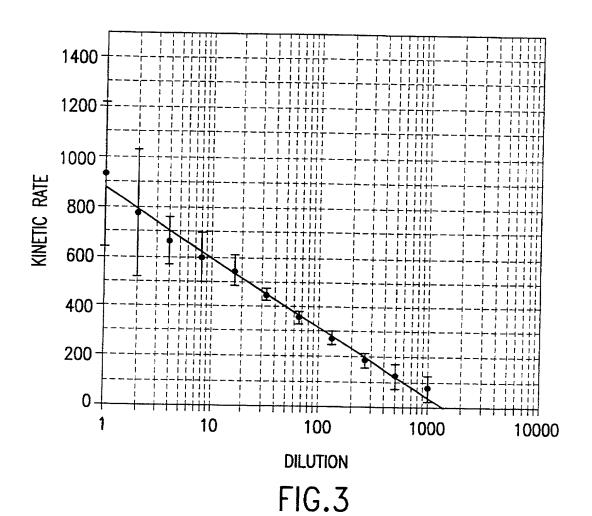
- f) detecting said label, wherein detection of said label is indicative of the presence of said first antibody.
- 17. A kit for the detection of a first antibody comprising a first container containing a bioassay plate having silver ions immobilized thereon.
- 5 18. The kit of Claim 17 further comprising a second container containing a biotinylated antigen that is reactive with said first antibody.
 - 19. The kit of Claim 18 further comprising a third container containing a labeled second antibody that binds to said first antibody.
 - 20. A kit for the detection of an antigen comprising a first container containing a bioassay plate having silver ions immobilized thereon.
 - 21. The kit of Claim 20 further comprising a second container containing a biotinylated antibody having specificity for said antigen.
 - 22. The kit of Claim 21 further comprising a third container containing an antibody having specificity for said antigen.
- 15 23. An apparatus for activating microplates comprising:
 - a) a housing;
 - b) a reagent addition/withdrawal chamber disposed in said housing, said reagent addition/withdrawal chamber including reagent and wash storage containers in communication with a manifold, said manifold in communication with dispense lines disposed to deliver wash and reagent to a microplate, and further including aspirate lines in communication with the manifold, said manifold in communication with a waste container, said aspirate lines disposed to aspirate spent reagent from said microplate;

- c) an incubation chamber disposed in said housing, said incubation chamber including a means for vertically delivering a non-reactive sealing plate to said microplate, and a means for heating and agitating said microplate.
- d) a means for horizontally conveying a microplate into and out of said addition/withdrawal chamber and between said addition/withdrawal chamber and said incubation chamber.

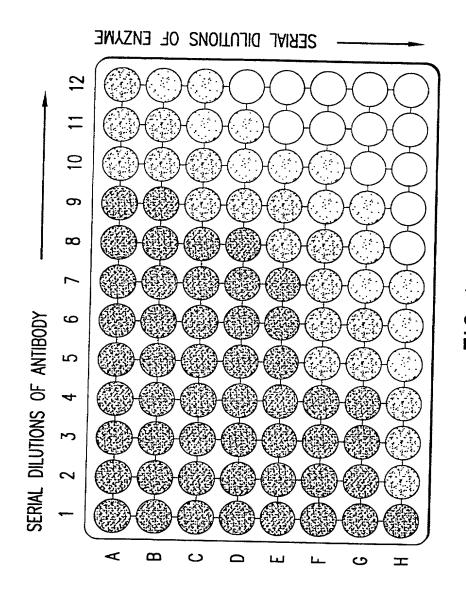




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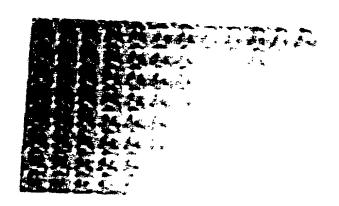
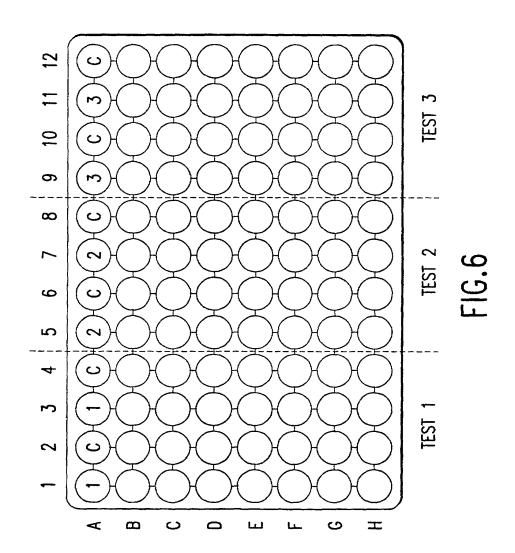
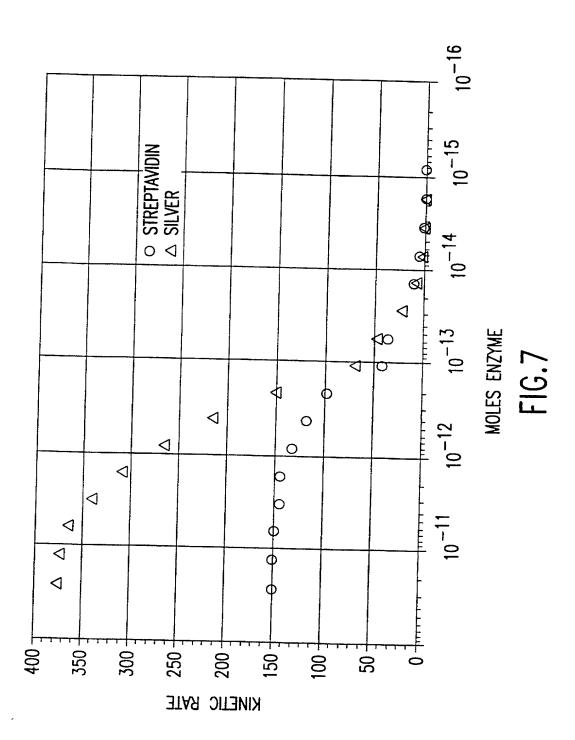
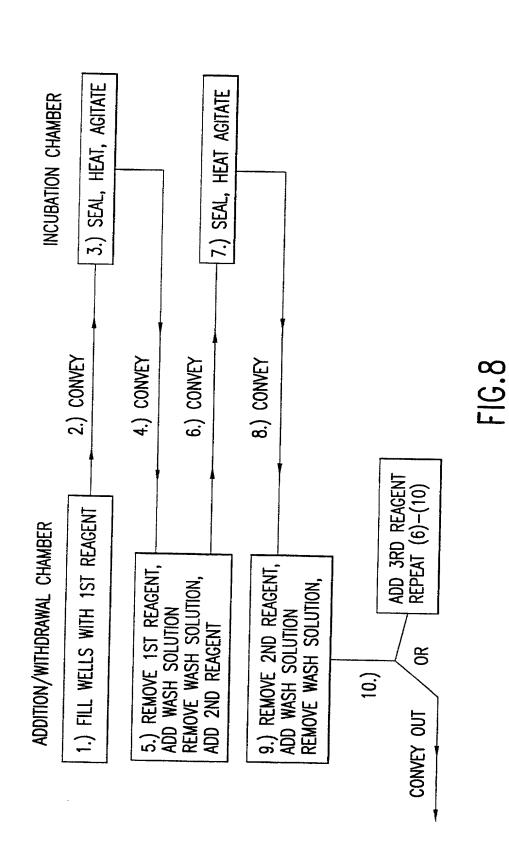


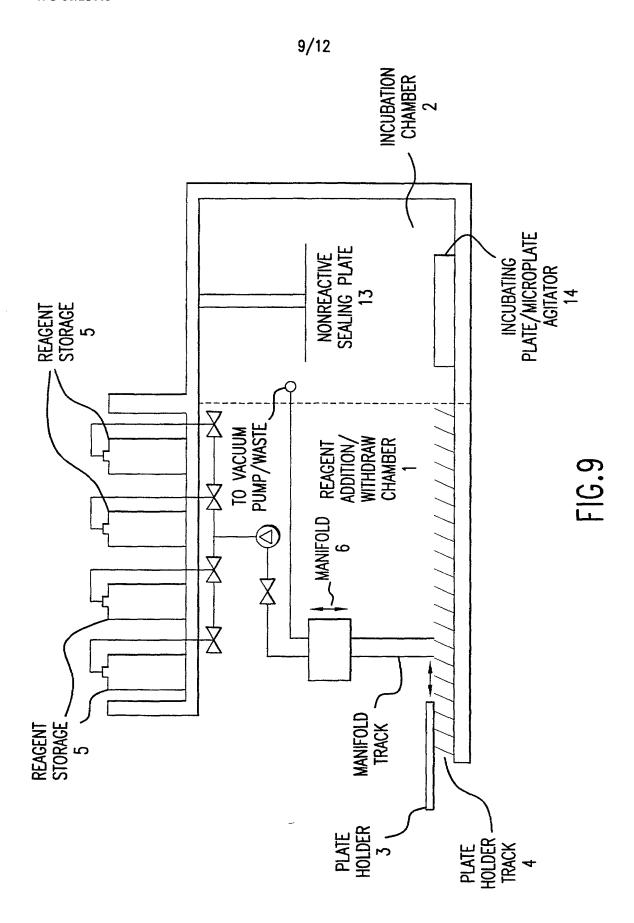
FIG.5

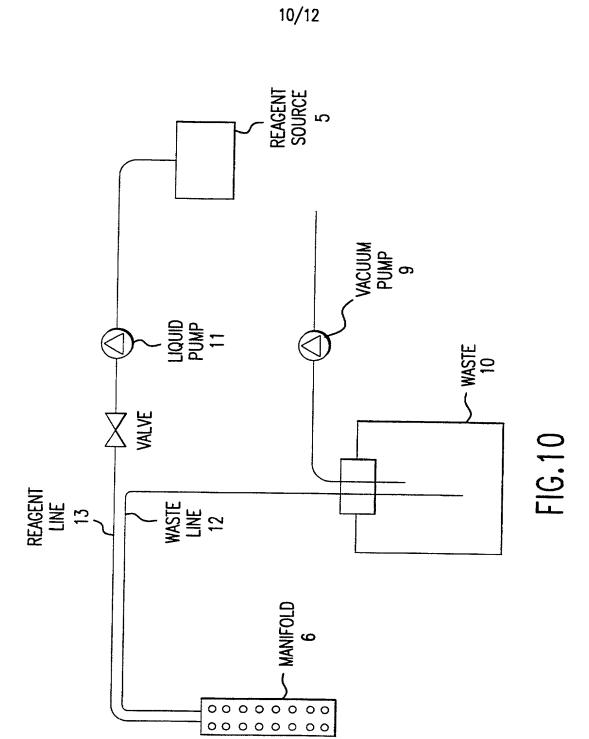


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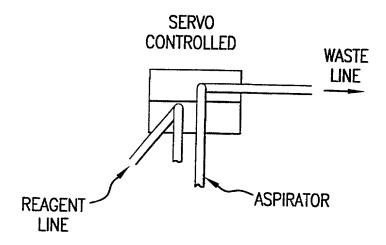


FIG.11A

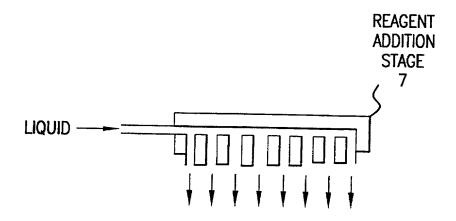
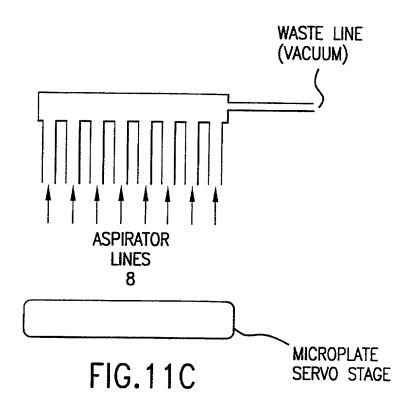


FIG.11B



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COMBINED DECLARATION AND POWER OF ATTORNEY

(Original, Design, National Stage of PCT, Divisional, Continuation or C-I-P Application)

As a below named inventor, I hereby declare that:

My residence, post office address and citizenship are as stated below next to my name; I believe I am the original, first and sole inventor (if only one name is listed below) or an original, first and joint inventor (if plural names are listed below) of the subject matter which is claimed and for which a patent is sought on the invention entitled:

IMMOBILIZED SILVER IMMUNOASSAY SYSTEM

This declaration is of the following type:
[] original
[] design
[X] national stage of PCT.
[] divisional
[] continuation
[] continuation-in-part (C-I-P)
The specification of which: (complete (a), (b), or (c))
(a) [] is attached hereto.
(b) [X] was filed on April 16, 2001 as Application Serial No. 09/807,663 and was amended on <i>(if applicable)</i> . (c) [X] was described and claimed in PCT International Application No. PCT/US99/23902 filed on October 14, 1999 and was amended on <i>(if applicable)</i> .
Acknowledgement of Review of Papers and Duty of Candor
Acknowledgement of Review of Papers and Duty of Candor I hereby state that I have reviewed and understand the contents of the above identified specification, including the claims, as amended by any amendment referred to above.
I acknowledge the duty to disclose information which is material to the patentability of the subject matter
claimed in this application in accordance with Title 37, Code of Federal Regulations § 1.56.
[] In compliance with this duty there is attached an information disclosure statement. 37 CFR 1.98.
Priority Claim
I hereby claim foreign priority benefits under Title 35, United States Code, § 119(a)-(d) of any foreign
application(s) for patent or inventor's certificate or of any PCT International Application(s) designating at least one

I hereby claim foreign priority benefits under Title 35, United States Code, § 119(a)-(d) of any foreign application(s) for patent or inventor's certificate or of any PCT International Application(s) designating at least one country other than the United States of America listed below and have also identified below any foreign application(s) for patent or inventor's certificate or any PCT International Application(s) designating at least one country other than the United States of America filed by me on the same subject matter having a filing date before that of the application on which priority is claimed

(complete (d) or (e))

- (d) [] no such applications have been filed.
- (e) [X] such applications have been filed as follows:



BAKER BOTTS L.L.P. FILE NO.: A32011-A-PCT-USA-072448.0313

COUNTRY	APPLICATION NO.	DATE OF FILING (day, month, year)	DATE OF ISSUE (day, month, year)	PRIORITY CLAIMED UNDER 35 USC 119
				[]YES NO []
		.		[]YES NO []
				[]YES NO []
LL FOREIGN API	PLICATION[S], IF ANY, FILED MORE THAN	12 MONTHS (6 MONTHS FOR DESIGN) PRIO	R TO SAID APPLICATION	
				[] YES NO []
				[] YES NO []
			İ	[] YES NO []

Claim for Benefit of Prior U.S. Provisional Application(s)

I hereby claim the benefit under Title 35, United States Code, § 119(e) of any United States provisional application(s) listed below:

Provisional Application Number	Filing Date			
60/104,263	14 October 1998			
60/145,786	27 July 1999			

Claim for Benefit of Earlier U.S./PCT Application(s) under 35 U.S.C. 120

(complete this part only if this is a divisional, continuation or C-I-P application)

I hereby claim the benefit under Title 35, United States Code, § 120 of any United States application(s) or PCT international application(s) designating the United States of America that is/are listed below and, insofar as the subject matter of each of the claims of this application is not disclosed in the prior application(s) in the manner provided by the first paragraph of Title 35, United States Code § 112, I acknowledge the duty to disclose information as defined in Title 37, Code of Federal Regulations, § 1.56 which occurred between the filing date of the prior application(s) and the national or PCT international filing date of this application:

Application Serial No.)	(Filing Date)	(Status) (patented, pending, abandoned)
<u> </u>		, , , , , , , , , , , , , , , , , , , ,
(Application Serial No.)	(Filing Date)	(Status) (patented, pending, abandoned)
	(Power of Attorney	
As a named inventor, I her	eby appoint Dana M. Raymond, Reg. No. 18,540; Frederi	ck C. Carver, Reg. No. 17.021; Francis J. Hone, Reg.

As a named inventor, I hereby appoint Dana M. Raymond, Reg. No. 18,540; Frederick C. Carver, Reg. No. 17,021; Francis J. Hone, Reg. No. 18,662; Joseph D. Garon, Reg. No. 20,420; Arthur S. Tenser, Reg. No. 18,839; Ronald B. Hildreth, Reg. No. 19,498; Thomas R. Nesbitt, Jr., Reg. No. 22,075; Robert Neuner, Reg. No. 24,316; Richard G. Berkley, Reg. No. 25,465; Richard S. Clark, Reg. No. 26,154; Bradley B. Geist, Reg. No. 27,551; James J. Maune, Reg. No. 26,946; John D. Murnane, Reg. No. 29,836; Henry Tang, Reg. No. 29,705; Robert C. Scheinfeld, Reg. No. 31,300; John A. Fogarty, Jr., Reg. No. 22,348; Louis S. Sorell, Reg. No. 32,439; Rochelle K. Seide Reg. No. 32,300; Gary M. Butter, Reg. No. 33,841; Marta E. Delsignore, Reg. No. 32,689; and Lisa B. Kole, Reg. No. 35,225 of the firm of BAKER BOTTS L.L.P., with offices at 30 Rockefeller Plaza, New York, New York 10112, as attorneys to prosecute this application and to transact all business in the Patent and Trademark Office connected therewith

SEND CORRESPONDENCE TO: BAKER BOTTS L.L.P. 30 ROCKEFELLER PLAZA, NEW YORK, N.Y. 10112 CUSTOMER NUMBER: 21003	DIRECT TELEPHONE CALLS TO: BAKER BOTTS L.L.P. (212) 705-5000
	<u></u>

I hereby declare that all statements made herein of my own knowledge are true and that all statements made on information and belief are believed to be true; and further that these statements were made with the knowledge that willful false statements and the like so made are punishable by fine or imprisonment, or both, under Section 1001 of



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FILE NO.: A32011-A-PCT-USA-072448.0313

Title 18 of the United States Code and that such willful false statements may jeopardize the validity of the application or any patent issued thereon.

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